General Chemistry Mr. MacGillivray Quiz #17: Ionic and Covalent Nomenclature

Name the following <u>ionic</u> compounds:			
1. CoCl ₃			
2.	CoCl ₂		
Give the formula for the following <u>ionic</u> compounds. 3. ammonium phosphate			
	· ,		
Name the following covalent compounds:			
5.	CCI ₄		
6.	Cl ₂ O ₇		
Give the formula for the following <u>covalent</u> compounds.			
7.	tetraphosphorus decoxide		
8.	nitrogen monoxide		

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Give the formula for the following covalent compounds.

7. tetraphosphorus decoxide

8. nitrogen monoxide

6. Cl₂O₇

Quiz #17: lonic and Covalent Nomenclature	77 1-2
Name the following ionic compounds: 1. CoCl ₃ 2. CoCl ₂ Cobalt (II) Chloride Cobalt (III) chloride	NH4 2003>
Give the formula for the following ionic compounds. 3. ammonium phosphate $(N+4)$, $PO4$ 4. nickel (II) sulfate $NiSO_{U}$	Ni+2Say
Name the following <u>covalent</u> compounds: 5. CC4 <u>Carbon</u> tetrachloride 6. Cl ₂ O ₇ <u>dichlarine</u> heptoxide	